

Biochemistry Exam Questions And Answers

M Walker

BI/CH421 Biochemistry I Exam 4 12/12/2016 - Boston University Part I: Multiple choice –Circle the choice that best answers the question 3 pts each. Which of the following statements about nucleic acids is false? Nucleic acids are polymers of nucleotides. Nucleic acids are linked together via phosphodiester bonds to the ribose or deoxyribose sugars. Nucleic acids are polyanions. Biochemistry Student Number: Winter 2018 Midterm Exam #3 Midterm Exam #3 Answer all questions on the test. Each multiple choice question has a value of two points and must be answered in pencil on the bubble sheet provided. The value for each short answer question is given with the questions. The final page of the exam has equations and other relevant information. Feel free to remove this page, *Enzymes: Practice Questions #1 - lecouredebias.com* Enzymes: Practice Questions #1 1. Compound X increases the rate of the reaction below. Compound X is most likely A. an enzyme B. a lipid molecule C. an indicator D. an ADP molecule 2. The equation below summarizes the process that produces the flashing light of a firefly. **Biochemistry Sample Items - National Board of Medical Examiners** In order to determine the rate of DNA synthesis in various mammalian tissues, the administration of which of the following labeled substances would be most specific in labeling DNA? 2. 2-year-old boy who recently emigrated from Somalia is brought to the physician because of a 1-day history of pain of his arms and legs. **This set of Biochemistry Multiple Choice Questions & Answers ...** This set of Biochemistry Multiple Choice Questions & Answers (MCQs) focuses on “Nucleic Acids”. Number of hydrogen bonds between adenine and thymine? Number of hydrogen bonds between guanine and cytosine? Which ratio is constant for DNA? Who discovered nucleic acid? Name the pyrimidine base which is found in RNA but not in DNA? *GRADUATE RECORD EXAMINATIONS Biochemistry, Cell and ...* 6 **BIOCHEMISTRY, CELL AND MOLECULAR BIOLOGY TEST PRACTICE BOOK** Test-Taking Strategies The questions in the practice test in this book illustrate the types of multiple-choice questions in the test. When you take the actual test, you will mark your answers on a separate machine-scorable answer sheet. **FRCPath in Clinical Biochemistry - Part 1 MCQ Paper Questions** The Part 1 comprises a single paper of 125 multiple choice questions (MCQs) in single best answer format set in broad topic areas aligned with the curriculum: • Laboratory management competencies **Question with Answers- Lipids - Florida Atlantic University** Question with Answers- Lipids A. The structures of different types of lipids are studied and compared. (Questions 1-8) 1. ____ Which of the following molecules is a typical fatty acid? a) A molecule that has an even number of carbon atoms in a branched chain. b) An

amphipathic dicarboxylic acid with unconjugated double bonds. **Biochemistry 2000 Sample Questions Amino Acids & Purification ...** Biochemistry 2000 Sample Questions Amino Acids & Purification (1) Give brief definitions or unique descriptions of the following terms: (a) oligopeptide (b) stationary phase (c) neutral drift (d) quaternary structure (e) zwitterion (f) salting out (2) Consider the 20 amino acids that commonly occur in proteins. Draw a stick diagram of Questions with Answers- Carbohydrates - Florida Atlantic University Which characteristic is shared by the ring forms of D-sorbose and D-galactose? Both contain a hemiacetal bond. Both exist mainly as furanoses. Both can undergo mutarotation. Both are stable at neutral pH. Which describes the relationship between D-sorbose and D-fructose? They are diastereomers that are also epimers. Chemistry/Biology 302 - Biochemistry: Exam 1 Practice Problems these questions come from several years of past Biochem tests. Over those years, we've used several different textbooks, and some of them use different approximate pKa values for buffers and amino acids. **QUESTION BANK SAQ - Lokmanya Tilak Municipal Medical ...** Structure and functions of three biochemically important disaccharides. Sugar derivatives. Explain the structure of starch and glycogen. 3. Lipid Chemistry. Define lipids and classify them with suitable examples. Classify fatty acids in various ways with ... **Practice ProblemsonCarbohydrates - uwo.ca** Carbohydrates.8 20. A student dissolved -D-galactose in acid, placed the solution in a sealed tube, and then monitored the optical activity of the solution in a polarimeter over time. **Questions with Answers- Proteins & Enzymes - Florida Atlantic University** Questions with Answers- Proteins & Enzymes A. A peptide with 12 amino acids has the following amino acid composition: 2 Met, 1 Tyr, 1 Trp, 2 Glu, 1 Lys, 1 Arg, 1 Thr, 1 Asn, 1 Ile, 1 Cys Reaction of the intact peptide with fluorodinitrobenzene followed ... Biochemistry Unit Exam - MIT OpenCourseWare Biochemistry Unit Exam Question 1 a) An example of a structural representation is shown in the adjacent box. Draw a structural representation of the amino acid, Aspartic acid, which has the side chain of: CH₂ COOH. b) The molecule drawn above is a monomer found in _____. c) Below is a pentose found in RNA. **Chapter 8An Introduction to Metabolism Multiple Choice Questions** Multiple-Choice Questions 1) Which term most precisely describes the cellular process of breaking down large molecules into smaller ones? A) catalysis The following are a selection of Clinical Biochemistry Part 1 exam ... The following are a selection of Clinical Biochemistry Part 1 exam questions retired from the College question bank in December 2020: Question 1 Serum from a man who has dermatomyositis had LDH isoenzymes analysed using thin-layer agarose gel electrophoresis. Which isoenzyme would be expected to have the highest peak? A. LD1 B. LD2 C. LD3 D. LD4 **FINAL EXAM MULTIPLE CHOICE - Boston University** Questions 1-17 are worth 10 points each (170 pts.) and questions 18-32 are worth 5 points each (75 pts.). 1. An enzyme reaction to be studied at pH 4.0 can best be carried out using a buffer solution made from which of the following acids, and their conjugate bases, with Kavalues as ... *Graduate Biochemistry Name: (Desai questions) questions ...* Graduate Biochemistry Name: Exam 3 Haas questions 1. Sudden infant death syndrome (SIDS) is a devastating disease that affects neonates within their first 6 months after birth. Typically, the child expires in the

early hours of the morning and is found dead in their crib with no underlying post-mortem autopsy indications. However, SIDS deaths are **Biochemistry 3300 Problems (and Solutions) Metabolism I** (15) Answer the following questions regarding glycolysis: (a) What is the overall chemical reaction for the 'preparatory' phase of glycolysis? (b) What is substrate-level phosphorylation?

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